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EFFECTS OF MINERAL SUBSTANCES WITH A BUFFERING EFFECT ON MILK PRODUCTION AND MILK COMPOSITION IN HEAT STRESS CONDITIONS

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This paper presents the results of investigations on the effects of a mineral mixture with a buffering effect, (bentonite, zeolite, magnesium oxide and sodium bicarbonate), on the quantity and composition of milk, pH value of cow's rumen contents, number and motility of microorganisms. The study was conducted from June to November on two groups of cows: the control - C (129) and experimental - E (119). The investigated mineral mixture was included in the complete feed mixture (1%). The milk yield during the hottest summer months (June - August) in the group E was higher by 1.99 kg or 8.16% (p<0.05). During the period from September to November the milk yield in this group was also higher, but insufficiently to make the difference (0.38 kg or 1.53%) between the C and E groups (p>0.05). The yield of milk was higher in the group E by 1.19 kg or 4.85% (p<0.05) for the duration of the experiment from June to November. The differences in the content and yield of milk fat and protein, (% and kg) were insignificantly higher for the group E during both investigation periods. The pH value of the rumen content was higher in the group E and was within the optimal physiological limits. Before the morning meal the values for the groups C and E were 6.25 and 6.79 (p>0.05), respectively. After the morning meal the corresponding rumen pH values were 6.01 and 6.92 (p<0.05). The number and motility of the infusoria present in the rumen of the group E cows was increased. The studied mineral mixture has a positive effect on the pH stability and development of the microorganisms present in the rumen under conditions of heat stress. This contributes to the arousal of favorite conditions for food digestion and utilization and hence to higher milk production and increased fat and protein content.

Key words: acidosis; cow; rumen; buffer

ЕФЕКТИТЕ ОД МИНЕРАЛНИТЕ МАТЕРИИ СО ПУФЕРЕН ЕФЕКТ ВРЗ ПРОИЗВОДСТВОТО НА МЛЕКО И НЕГОВИОТ СОСТАВ ВО УСЛОВИ НА ТОПЛОТЕН СТРЕС

Во овој труд се презентирани резултатите од истражувањето за ефектите на минерални смеси со пуферен ефект (бентонит, зеолит, магнезиумоксид, натриумбикарбонат), врз количеството и составот на млекото, pH-вредноста на содржината на руменот кај крави, како и врз бројот и подвижноста на микроорганизмите. Истражувањето беше спроведено од јуни до ноември на две групи крави: контролна - С (129), и експериментална – Е (119). Истражуваната минерална смеса беше додадена во комплетната гранителна смеса (1%). Приносот на млеко во текот на најжешките летни месеци (јуни-август) во групата Е беше повисок за 1,99 kg (8.16%) (p<0,05). Во текот на периодот септември-ноември приносот на млеко во оваа група исто така беше повисок, но недоволен за да се направи разлика (0,38 kg, или 1,53%) помеѓу групите С и Е (p>0,05). Приносот на млеко во групата Е беше повисок за 1,19 kg, или 4,85% (p<0,05) за цело време на траењето на експериментот (јуни-ноември). Разликите во содржината и приносот на млечни масти и протеини (во % и kg) беа незначително повисоки во групата Е во текот на двата истражувачки периода. Вредноста на рН на содржината на руменот беше повисока во групата Е, но во рамките на оптималните физиолошки граници. Пред утринскиот оброк вредностите за групите С и Е беа 6,25 и 6,79 (p>0,05), соодветно. По утринскиот оброк соодветните pH-вредности на руменот беа 6,01 и 6,92 (p<0,05). Бројот и подвижноста на инфузории во руменот на кравите од групата Е се зголеми. Истражуваната минерална смеса има позитивен ефект врз стабилноста на pH и развојот на микроорганизмите присутни во руменот во услови на топлотен стрес. Ова придонесува за поттикнување погодни услови за варење и користење на храната, а тоа значи повисоко производство на млеко и зголемување на содржината на масти и протеини во него.

Клучни зборови: ацидоза; крава; румен; визуелен ефект

INTRODUCTION

Disorders in the process of food digestion in the rumen often manifest as acidosis which leads to a number of pathological processes among which are morphological changes of the rumen mucosa, decreased motility, poor food passage, as well as ruminitis and parakeratosis. As a result of the above mentioned is a significantly decreased resorptive surface of the rumen and reduced absorption of the products of food digestion. Their hindered input can jeopardize the metabolic balance, above all the stability of energy metabolism. Bearing in mind that acidosis is accompanied by poor appetite it is clear that the end result of these disorders is a reduction in milk production.

Rumen acidosis is characterized by an excessive accumulation of lactic acid as the result of a diet rich in easily digested carbohydrates, excessive acidic and moist feeds, insufficient fibers in the diet, few meals and long periods between meals. In such cases there is insufficient production of saliva, which is the natural regulator of the electrochemical reaction in the rumen. These problems are specially enhanced in high yielding dairy cows in the first phase of lactation when they have to consume large quantities of concentrated feeds and in the summer months, especially when kept in tie stalls when high ambient temperatures and inadequate microclimate in the stable result sudden appetite loss. Besides, during the high summer temperatures food is at a higher risk to become spoiled, silage ferments, moulds which produce mycotoxins are produced and rumen microflora is negatively affected which makes the problem even worse.In the state of acidosis the number of bacteria producing lactic acid in the rumen progressively increase, and the number of bacteria which use up this acid is on the decrease. Streptococcus bovis produces lactic acid in the process of carbohydrate metabolism and guickly becomes the dominant microorganism in the rumen's content.During this process the number of Megasphera elsdenii, Selenomonas ruminantium and other bacteria which use lactic acid is on the decrease. By further lowering the electrochemical reaction the reproduction of Streptococcus bovis decreases and lactobacilli continue to produce lactic acid thus contributing to a further increase in its concentration in the rumen. The lowest pH values in the rumen are recorded during night time and in the early morning hours. The deviation of the pH value from its normal physiological range (6.26.8) results in an adverse impact on the development of the rumen microflora and thus the processes of digestion, which negatively affects the production and fat content in the milk (Horea et al., 2006)

The key factor for the preservation of the pH balance of the rumen is the bicarbonate buffer. However, under conditions of rumen acidosis there is a decreased production of saliva and hence less bicarbonate reaches the rumen. The efficacy of the bicarbonate is then almost negligible due to the present atonia of the rumen and insufficient mixing of the saliva and rumen content. Due to the state of atonia there is a decreased degree of resorption of acids and the role of the bicarbonate buffer is almost minimized. Under conditions of heat stress, during high summer temperatures (>27°C) due to panting there is increased Salivation which results in a loss of calcium, potassium, sodium and magnesium cations which are necessary for the maintenance of the acid-base status of the animal (Vujanac et al., 2011).

Buffers which have the capability to neutralize the increased acidity are used for the maintenance of the rumen's pH. Some minerals, such are bentonite and zeolite belong to the group of natural alumosilicates which have a marked surface activity which is the result of superficial electricity. Their chemical activity can be controlled and directed, which gives them the chance to be applied in the diet of domestic animals. Both of these minerals act as buffers. In an acidic environment by cation exchange they bind H⁺ and release a cation which is naturally found in an exchangeable position, thus resulting in increased pH. These changes occur only as long as the neutral environment (pH 7) is achieved. When the pH is over pH 7, the presence of these minerals leads to a decrease in pH down to the neutral stage. Rivera et al. (2002) have confirmed that zeolite has a tendency to neutralize the water solutions, regardless if it serves as a donor or acceptor of protons, which indicates its amphoteric character. The use of only magnesium oxide or sodium bicarbonate can result in an increased rumen pH which is not desirable. After their reduction in the rumen there is a subsequent pH decrease below optimal values. The application of a mixture containing bentonite, zeolite, magnesium oxide and sodium bicarbonate creates conditions for the maintenance of a stable pH in the rumen within the optimal range for a period of 24 hours. Besides this, these substances show other useful effects. Magnesium oxide enhances the absorption of acetic acid. Bentonite and zeolite successfully absorb micotoxins (aflatoxin), excess ammonia, heavy metals, radio nucleotides and excess water. Bentonite swells and thus slows food passage through the digestive tract which results in its improved digestion and utilization. Such additives are added to the feed mixture in a quantity of 1 - 2%. (Garicia Lopez, 1992; Vicini et al., 1988; Eng et al., 2002; Nikkah et al., 2000; Adamović et al., 2003; Petrujkić et al., 2010; Radivojević et al., 2010 and others).

MATERIAL AND METHODS

The study was conducted from June to November on two groups of cows: the control - C (129) and experimental -E (119). The experiment was carried out on the Padinska Skela property which is part of the PKB Corporation. The cows were in a linked system of keeping. Groups were formed according to "couple" system (two cows with similar characteristics) after 7-10 days calving and alloted to one of two stables. The given meals were identical with the difference that in the complete feed mixture in the group E a mineral supplement with buffering propreties was added in the quantity of 1%. The mixture was produced at the Institute for Technology of Nuclear and Other Mineral Raw Materials, Belgrade, Serbia (commercial name Mix Plus). The mixture composition is based on magnesium oxide, sodium bicarbonate, bentonite and zeolite. Cows were fed twice a day (6-7 AM and 5-6 PM). Feedstuff composition based on the proportion of dry matter of individual feeds in dry matter of the given meal, chemical composition of the meal, and composition of complete feeds are given in Tables 1-3. Extruded soybeans were given separately, an average of 1.5 kg/cow/day. Concentrate was standardized to the amount of 300 g/kg of milk. Control of milk yield and determination of the chemical composition of milk was carried out once a month. The balance of cations and anions in the meals (BAKO) was determined using the formula proposed by (Beed, 1992):

BKAO mEq/kg SM = $[(435 \times \%Na) + (256 \times \%K)] - [(282 \times \%Cl) + (624 \times \%S)]$

Feed consumption was estimated daily. The pH of the rumen content was measured at the end of the trial. Samples of ruminal contents were collected by probe 0.5 h before morning feeding and

5 h after the morning feeding from 10 cows in each group. The electrochemical reaction in the samples of rumen contents was measured by a pH meter (WTW 330i) immediately after sampling. Chemical analysis of the milk samples was performed with an automatic machine Milkoscan.

The air temperature was measured daily (1-2 PM) with a mercury thermometer. Outdoor temperature was measured in the shade at 10 meters distance from the stables at a height of 150 cm from the soil, and the stable temperature was recorded in the middle of the building, at 100 cm above the feeding table.

Table 1

Percentage of dry matter in dry matter of complete feed meal, %

Indicator	C group	E group
Alfalfa hay	17.35	16.59
Alfalfa haylage	11.09	10.90
Corn plant silage, 30% DM	28.61	29.71
Brewer's trope	3.73	3.57
Extruded soybean	6.81	6.51
Complete feed mixture C	32.41	-
Complete feed mixture E	-	32.72
Total	100.00	100.00

Table 2

Chemical composition and cation and anion content in the meals

Indicator	C group	E group
Dry matter in the meal,%	49.55	49.24
Dry matter, kg	19.82	20.73
NEL, MJ	127.58	132.89
Proteins,% DM	16.65	16.78
Fats,% DM	4.43	4.39
Cellulose, % DM	18.06	17.85
Ca, % DM	0.89	0.91
P, % DM	0.66	0.65
K,% DM	1.50	1.48
Na,% DM	0.13	0.15
Mg,% DM	0.34	0.39
Cl,%DM	0.45	0.44
S,% DM	0.30	0.29
BAKO mEq g/DM	+126.45	+139.09

Table 3

Indicator	C group	E group		
Corn grain groats	26.30	26.30		
Oat grain groats	20	20		
Sunflower meal	42	42		
Flour meal	8	7		
Dicalcium phosphate	1.40	1.40		
Chalk	1.60	1.60		
Vitamin & mineral mixture	0.70	0.70		
Mix Plus	0	1		
Total	100	100		
NEL, MJ	5.98	5.92		
Total protein	18.68	18.54		

Composition of complete feed mixtures

Statistical data analysis was done with StatSoft. Inc.: Statistica. Data Analysis Software System. Version 8.0, 2008.

RESULTS AND DISCUSSION

Differences between the diets for the groups C and E during the experiment were negligible and it can be stated that they were isoenergetic and isoproteinic. The difference was evident for the concentrations of anions and cations (BKAO) which was higher for the group E and measured for C:E 126,45:139.09 mEq g/DM. The value for the group E was at the same time closer to the optimal BKAO value for mid-lactation cows (100–200 days) included in both groups C and E.

Previous studies confirmed that consumption of DM, water and milk production increase as BKAO increases in the meals for lactating cows. BKAO values in the experimental group in the described trial are close to the values reported by Delaquis and Block (1995); Tucker and al. (1988), West (1993) and others which are considered acceptable for the mid lactation period (+150 to +250 mEq g/DM). According to the above authors maintenance of the high anion-cation difference, with higher cation values is necessary as Na⁺ and K^+ to prevent pH decrease and acidosis. Cows fed meals with a positive BKAO (+200 mEq/kg SM) produced more milk than cows fed a diet with a negative BKAO (-100 mEq/kg SM). pH values of the blood and urine, as well as HCO₃⁻ concentrations linearly increased with increasing BKAO values.

The milk yield in the period from June to August (Table 4) in the group E was higher by 1.99

kg or 8.16% (p<0.05). For the same period a relative (%) and absolute (kg) increase in the fat content and produced fats and proteins in the milk (p>0.05) was recorded. The average outdoors air temperature between 1–2 PM was 32° C and the temperature in the stables was 36° C

Table 4

Milk yield and composition of milk during the period from June to August

C group	E group	C = 100
24.36	26.35*	108.16
3.31	3.40	102.71
0.81	0.89	109.87
2.80	2.84	101.43
0.68	0.75	110.29
	24.36 3.31 0.81 2.80	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

*(p<0.05).

The milk yield in the period from September to October in the group E (Table 5) was higher by 0.38 kg, or 1.53% (0<0.05). Differences between the groups C and E were significantly smaller. The quantity of produced milk fats and proteins in this period was, as in June – August insignificantly higher in the group E (p>0.05). The average outdoor air temperature for the period between 1–2 PM was significantly lower compared to the previous period and measured 21°C, the stable air temperature was 25°C.

Table 5

Milk yield and composition of milk during the period from September to November

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Indicator	C group	E group	C = 100
Milk yield, kg/cow/day	24.76	25.14	101.53
Milk fat, %	3.33	3.49	104.80
Milk fat, kg/day	0.82	0.88	107.32
Proteins, %	2.92	2.93	100.34
Proteins, kg/day	0.72	0.74	102.77

The milk yield during the period from June to November, and the entire trial period (Table 6)

was higher in the group E by 1.19 kg, or 4.85% (p<0.05). The quantity of produced milk fat and protein was not significantly higher in the group E (p>0.05).

Table 6

Milk yield and composition of milk during the period from June to November

Indicator	C group	E group	C = 100
Milk yield, kg/cow/day	24.56	25.75	104.85
Milk fat, %	3.32	3.45	103.91
Milk fat, kg/day	0.81	0.89	109.88
Proteins, %	2.86	2.89	101.05
Proteins, kg/day	0.70	0.74	105.71

Comparing the results achieved in the produced quantity of milk in the period from June to August and September to November it can be concluded that the studied mineral mixture (with buffering action) effectively influenced the milk yield of cows in the warmest period of the year, when the cows were apparently exposed to the impact of heat stress as evidenced by temperatures which were above 27°C for a longer period of time.

Nikkhah et al. (2001), found that the use of zeolite and sodium bicarbonate in the diet of cows individually or in combination, increases the milk yield by 0.7-2 kg (p <0.05), while fat and protein in the milk increased by 0.12 to 0.26, respectively, equivalent to 0.04 to 0.06 percentage points. According to Adamović et al. (2003), the addition of mineral mixtures based on magnesium oxide, sodium bicarbonate, bentonite and organozeolite (modified natural zeolite) in the first 100 days of lactation cows influenced the increase of milk yield, corrected to 4% milk fat, by 0.43 kg/day or 1.59%. The average milk fat content in milk was higher in the experimental group by 0.29 percentage points (3.29 and 3.58%). Horea et al. (2006) when using the same mineral mixture as in the present study, found milk production to be increased by 1.5 kg/day or 6.17% with an increase in the milk fat content from 3.39 to 3.53%, and a slight increase of protein from 2.88 to 2.89%. Similar results in terms of milk yield when using mineral mixtures of the same composition were recorded by Petrujkić et al. (2010). In doing so, the differ-

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ences in the quantity of milk between the control and experimental groups of cows (30th, 60th and 90th day of the trial) were statistically significant (p<0.05). Bergero et al. (1995) found that the addition of 250/head/day of zeolite (clinoptilolite 64%) in the diets of cows increased the amount of milk by 1.47 kg/day or 12.47%, but did not have a major influence on the pH of the rumen contents, the amount of ammonia and content of volatile fatty acids present in the rumen. The results given by Moghaddam i Taghizadeh (2001) should be emphasized as they studied the effects of 1.8% and 3% zeolite in the meal DM during a 21 day trial on dairy cows. The authors determined a significant decrease in rumen pH (from 6.78 to 6.71 and 6.34). Results of the effects of mineral mixture with buffering effect (of the same composition, with the same rations and cows, as in our research,) on the pH of the rumen contents of cows were reported by Horea et al. (2006) (Table 7).

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Rumen pH values

Time of sampling	C group	E group	Significance
0.5 h before feeding	6.25 ± 1.5	$6.79\pm0.89*$	p<0.05
5 h after feeding	6.01 ± 2.8	$6.92\pm1.01*$	p>0.05

According to the research done by the authors the investigated use of mineral mixture had a beneficial effect on the pH of the rumen contents before (6.79) and after the morning feeding (6.92). This clearly shows that the presence of mineral mixture in the rumen contents maintained a stable electrochemical reaction continuously within the physiological range. Also in this group of cows there were small individual differences, which was not the case with cows in the group C, where the pH of the rumen content was below the physiological limits and was 6.25 before feeding and 6.01 after feeding. In such cases the animals did not take enough food in the morning and usually large amounts of food were left over up to the next meal. Similar results, when using the same mineral mixture in the diet of cows in early lactation, were found by Adamović et al. (2003) and Vujanac et al. (2005). Nikkhah et al. (2001) found that zeolite and sodium bicarbonate, administered simultaneously, increased the pH in the rumen liquor from 6.16 to 6.50. Similar results were determined by Petrujkić et al. (2010) as well. The pH of the rumen contents of the control and experimental groups of cows on the 60^{th} day was 6.22:6.77 and on the 90th 6.31:6.68, whereas the differences were statistically significant (p<0.05).

In addition to qualitative and quantitative changes in the bacterial flora in the rumen, the low pH affects the abundance and mobility of infusoria in the rumen (Nocek, 1997). In the study conducted on the native rumen contents significant differences between the presence of infusoria between the group C and E were found. Cows in the group C had significantly fewer infusoria in the visual field small and medium-sized (n = 8), and the number of large infusoria in the visual field was very small (1 to 2). In the group E of cows three times more infusoria in the visual field (24) was found and all three types were present: small, medium and large. Unlike cows in the group C, where the infusoria of the rumen contents were poorly motile, cows infusoria in group E were very mobile. Radivojević et al. (2010) used a similar mineral mixture in the experimental group of cows and found a higher pH value of rumen contents and a number of medium and large infusoria.

CONCLUSION

The quantity of milk in the hottest summer months (June – August) in the group E was higher by 1.99 kg or 8.16% (p<0.05). In the period from September to November the quantity of milk the in group E also increased, but differences between groups of 0.38 kg or 1.53% in favor of group E were lower (p>0.05). The quantity milk for the period from June to November was higher in the group E by 1.19 kg or 4.85% (p<0.05). In the same period, the differences in content and amount of milk fat and protein were slightly higher in the group E (p>0.05).

The pH of the rumen was higher in the group E, and before the morning feeding it was 6.25 and 6.79 (p>0.05), or after the morning feeding 6.01 and 6.92 (p<0.05). In the rumen in the experimental group an increased number of infusoria with greater mobility was described. It can be concluded that the investigated mineral mixture has a positive impact on the stability and the pH of the rumen microbial growth, which contributes to the

creation of favorable conditions for the digestion and feed utilization, and thus greater milk yield and production of milk fat and protein. This effect is more favorable when the animals are under conditions of heat stress caused by high external temperatures, reaching a value of over 27°C.

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